



## Certification Boulevard

### *Test Your Knowledge of Disinfection • Answer Key*

1. What happens to the chlorine residual when the nitrite (NO<sub>2</sub>) content in the effluent increases?
  - A. The residual stays the same ... nitrite has nothing to do with chlorine demand
  - B. The residual goes down**
  - C. The residual goes up
  - D. The residual doubles

*Nitrite has a demand for chlorine equal to about 5 lbs of chlorine for every 1 lb of nitrite to be oxidized. As nitrites go up, chlorine residual goes down.*

2. If your plant has a flow rate of 2.75 mgd, a chlorine demand in pre-treatment of 2.5 mg/L, pre-filtration of 1.5 mg/L and final effluent of 6.5 mg/L, and you maintain a chlorine residual of 2.5 mg/L, how many lbs/day of chlorine will be used?
  - A. 241 lbs/day
  - B. 1,398 lbs/day
  - C. 298 lbs/day**
  - D. 57 lbs/day

$$\begin{aligned} \text{Total mg/L} &= 2.5 + 1.5 + 6.5 + 2.5 = 13 \text{ mg/L} \\ \text{Lbs/day} &= \text{flow, mgd} \times \text{conc., mg/L} \times 8.34 \\ &= 2.75 \text{ mgd} \times 13 \text{ mg/L} \times 8.34 \\ &= 298.1 \text{ lbs/day} \end{aligned}$$

3. What chemical is used to identify a chlorine leak?
  - A. Ammonia**
  - B. Sulfur Dioxide
  - C. Sodium Hydroxide
  - D. Sulfuric Acid

*Ammonia fumes should be used to identify leaks ... do not spray liquid ammonia directly on fittings or valves.*

4. One volume of chlorine liquid will expand 92.84 times when it evaporates to a gas?  
*True or False*

*False: While it is true that chlorine liquid expands when converted to a gas, it actually expands nearly 460 times the original volume.*

5. Which effluent quality condition may cause the most problems with efficiency of a UV disinfection process?

- A. High NH<sub>3</sub>
- B. High NO<sub>3</sub>
- C. High NO<sub>2</sub>
- D. **High TSS**

6. Calculate the chlorine demand, given the following data:

- Total daily pounds used = 240 lbs/day
- The plant flow = 2.75 mgd
- The effluent chlorine residual = 2.5 mg/L

- A. 158 lbs/day
- B. 240 lbs/day
- C. 297 lbs/day
- D. **182 lbs/day**

*Supply – Residual = Demand*

- *Supply is given at 240 lbs/day*
- *Residual = 2.75 mgd x 2.5 mg/l x 8.34 lbs/gal = 57.3 lbs/day*
- *Demand = 240 lbs/day – 57.34 lbs/day = 182.7 lbs/day*

7. What action should never be performed when working with liquid chlorine?

- A. Wear a respirator
- B. **Trap liquid between two closed valves**
- C. Feed liquid to an evaporator
- D. Use ammonia to test for leaks

8. What does this formula best represent?

$$\frac{\text{Tank Volume, ft}^3}{\text{Flow, mgd} \times 92.84 \text{ cfm/mgd}}$$

- A. Chlorine residual
- B. **Contact chamber detention time, minutes**
- C. Contact chamber detention time, hours
- D. Tank volume in gallons

9. Leaking chlorine liquid will tend to collect near the floor of a closed room.

**True** or False

*True: Chlorine liquid is about 1.5 times heavier than water and, during a leaking condition, will typically remain on the floor. As the liquid evaporates to a gas, the gas will also remain low lying – gas is about 2.5 times heavier than air.*

10. What type of chlorine residual is created when ammonia is present and breakpoint has not been accomplished?

- A. **Combined**
- B. Free residual
- C. Tri-chloride
- D. Dioxide

*Combined residual (chloramines) predominates before the breakpoint.*

*Please forward your comments and sample questions for publication to:*

*Roy Pelletier, Assistant Bureau Chief  
City of Orlando Public Works Department  
Wastewater Bureau  
5100 L.B. McLeod Road  
Orlando, Florida 32811*

[roy.pelletier@cityoforlando.net](mailto:roy.pelletier@cityoforlando.net) (407) 246-2213